



*Institut for Fysik, Kemi og Farmaci*

# Battery (related) research at SDU

**Per Morgen**

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in collaboration with

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and

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# Activities at Odense, 1973-2013

## Surface physics

- Physics (electronics and optics) of clean surfaces
- Reactions and chemistry at surfaces
- Thin film systems: Oxides, nitrides, carbides and metals
- Tribology and rheology
- Analytical techniques

## Materials science

- Nanostructured materials: 1D-, 2D-, and 3D SiC; porous alumina
- Plasmonic structures: Au, Ag, and Cu on inverted nanodomes (anod. alumina)
- Electrode- and electrolyte materials and deposition for microbatteries (Bangalore)

# Energy and environment

## Vacuum based processing

- Thin film composite systems (SONOS) for flash disks; microbatteries (plasma))
- New dielectrics for MOS
- Ultrathin SiC films on Si
- In-situ growth of oxides, carbides and nitrides (films)

## Furnace or wet processes

- Conversion of C into SiC (hydrogen storage; diesel particle filters and catalyst supports; fuel cell catalyst support)
- Anodization of Al forming self-organized porous alumina
- Li-air batteries
- Corrosion studies

# Energy R/D in Odense

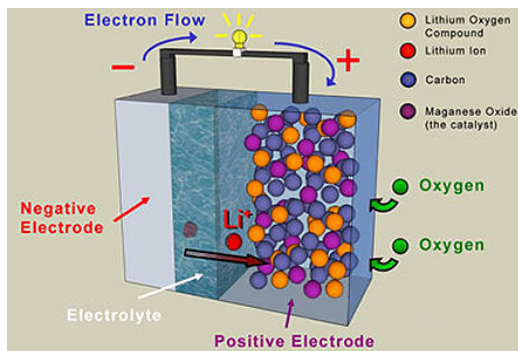
## History at SDU (Odense)

- Energilaboratoriet (Johs Jensen et al.): Li-batteries
- Development of electrolyte with Harwell (Li-ion)
- Spin-out companies: Hope Batteries, IRD, Danionics, and one-man companies
- Technical Faculty (2012): An education in Energy Systems

## Present developments

- Fuel cell research with IRD
- Functional nanomaterials with Kiel (Fachhochschule)
- Hydrogen storage materials
- Photovoltaic nanostructures
- ReLiable: Li-air batteries
- Supercapacitors with Kiel

# ReLiab Odense



Post. doc. Rajnish Dhiman, PhD

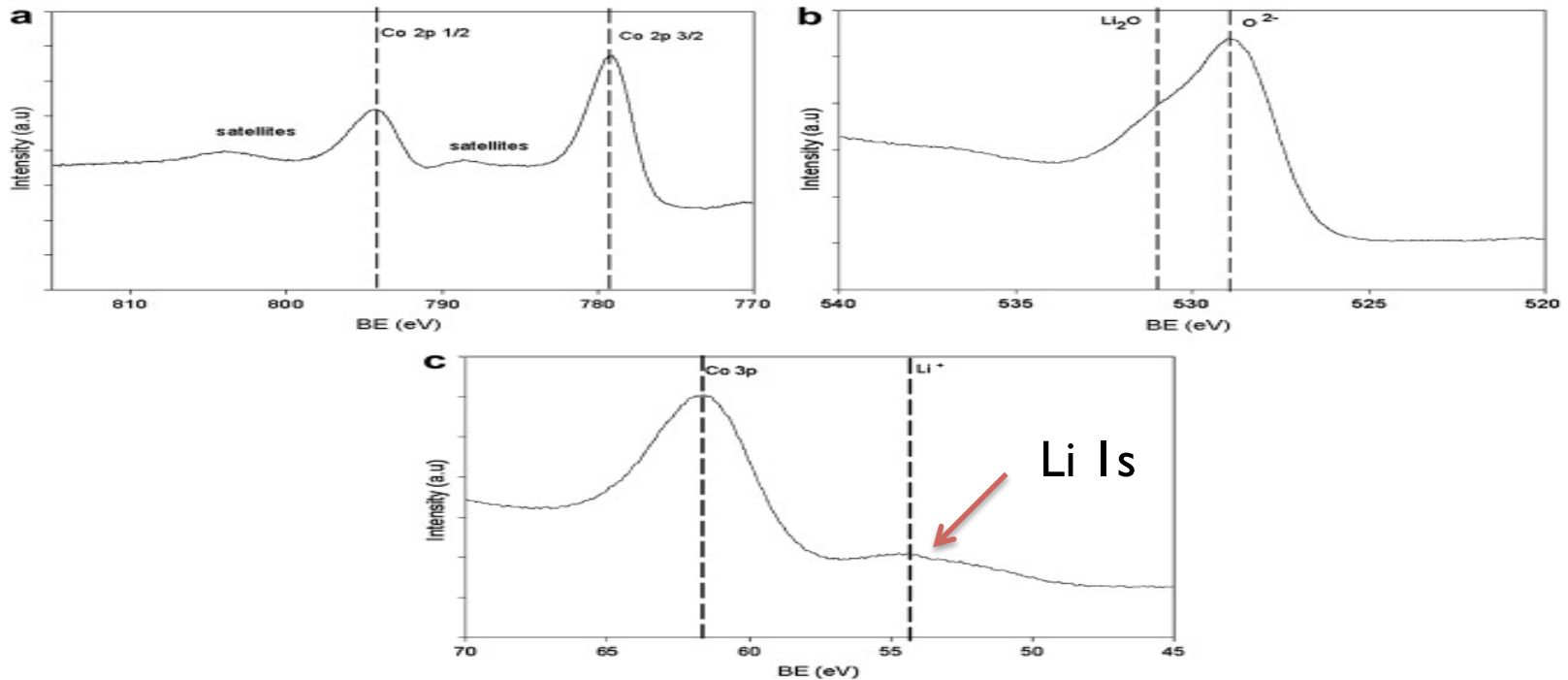
Work packages:

- (1) Electrolyte selection (DME for now)
- (2) Studies of interfaces with relevant techniques (from prototypes)

Facilities:

- Glove box and transfer system
- Surface analytical instrumentation
- AFM and STM
- Vacuum based deposition and plasma
- Raman microscopes
- X-ray diffraction
- Electrochemical workstations (Tech.)
- Electron microscopes with EDS / AES
- Gas evolution monitoring

# Examples of XPS ( $\text{LiCoO}_2$ )



**Co 2p, O 1s and Li 1s XPS spectra of  $\text{LiCoO}_2$  thin film.**

C.S. Nimisha , M. Ganapathi , N. Munichandraiah , G. Mohan Rao  
Studies on the target conditioning for deposition of  $\text{LiCoO}_2$  films

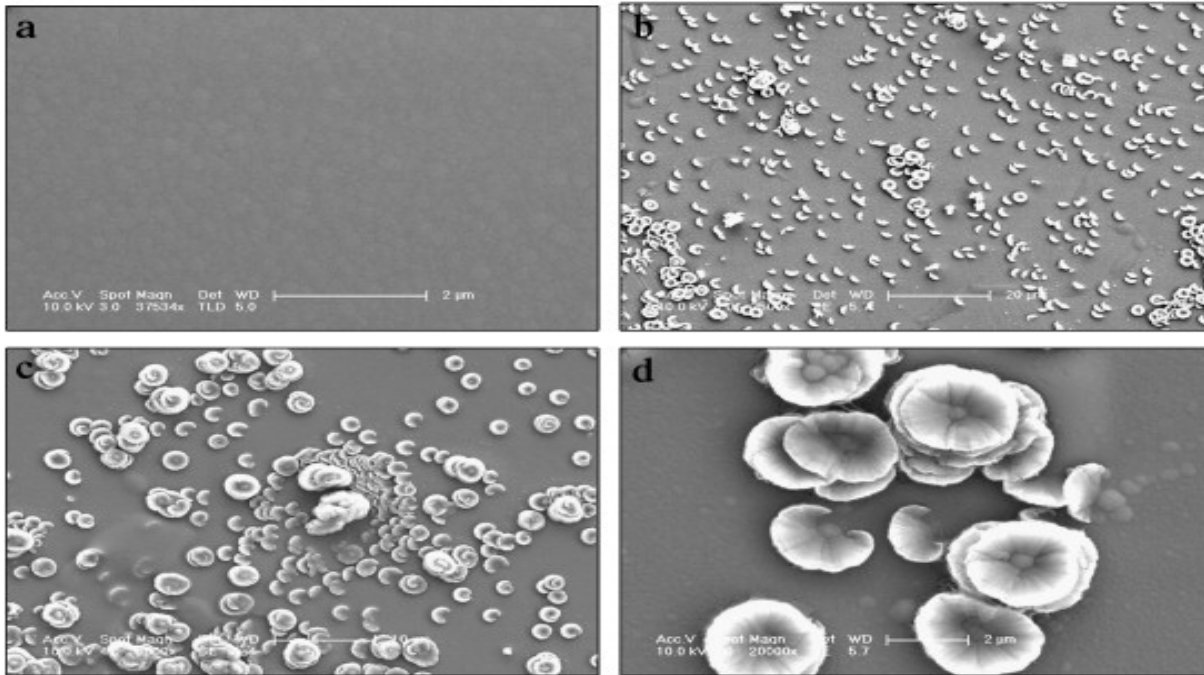
Vacuum Volume 83, Issue 6 2009 1001 - 1006

<http://dx.doi.org/10.1016/j.vacuum.2008.12.002>

# Other techniques

SEM

LiPON  
sputtered film



**SEM micrographs from LiPON films, (a) fresh sample after fabrication, (b), ( c ), (d) exposed samples**

C.S. Nimisha , G. Mohan Rao , N. Munichandraiah , Gomathi Natarajan , David C. Cameron

Chemical and microstructural modifications in LiPON thin films exposed to atmospheric humidity

Solid State Ionics Volume 185, Issue 1 2011 47 - 51

<http://dx.doi.org/10.1016/j.ssi.2011.01.001>

# Materials issues for Li-batteries

- Electrode materials (ex. non-Li anode materials)
- Electrolyte materials (aprotic for Li-air)
- Packaging (sealing, temperature control)
- Connections (stable, non-corrosive)
- Charging and discharging process reversibility
- Role of material crystallinity and crystallography (perovskites)
- New nanostructures to overcome lacking functionalities of conventional material classes (high porosity)
- Tailoring of material conductivity and stability vs. cycling
- New combinations of materials (ionic and electronic cond.)



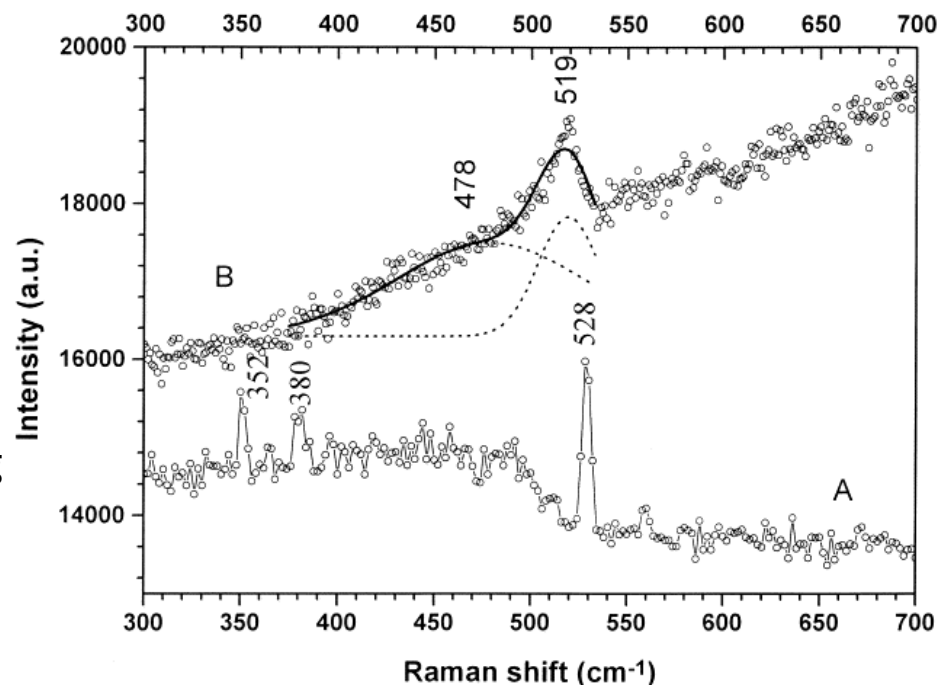
# Working with Li on Si (SiC)

The crystal structure and morphology of nanosized Si particles and wires after Li-insertion/extraction electrochemically have been studied by ex-situ XRD, Raman spectroscopy and electronic microscopy.

Raman spectra of the Li-inserted SiNWs electrode after annealing treatment.

(A) Li insertion;

(B) after step (A), then vacuum-annealing at 400°C for 5 h.

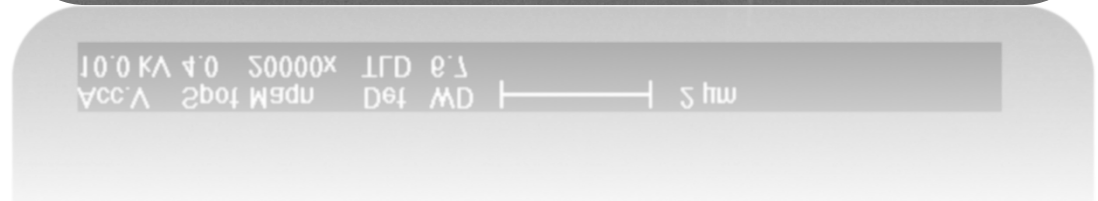
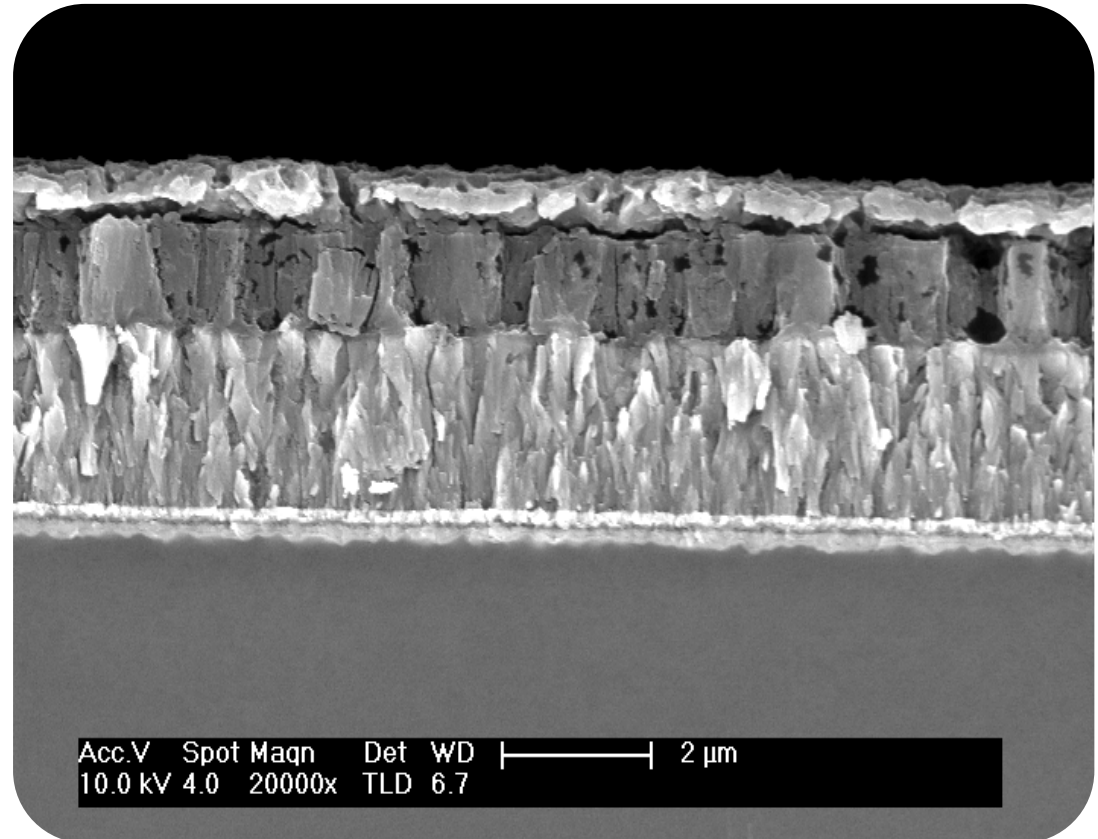
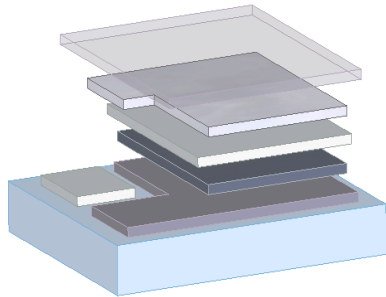


**The crystal structural evolution of nano-Si anode caused by lithium insertion and extraction at room temperature (Solid State Ionics)**  
Hong Li<sup>a</sup>, Xuejie Huang<sup>a</sup>, Liqian Chen<sup>a, \*</sup>, Guangwen Zhou<sup>b</sup>, Ze Zhang<sup>b</sup>, Dapeng Yu<sup>c</sup>, Yu Jun Mo<sup>d</sup>, Ning Pei<sup>d</sup>

# Li “studies” in Odense for Li-air

- Li adsorbed on Si surfaces (with SAES getter sources):
  - Thermal treatment, diffusion, desorption, structural changes of surface or subsurface regions
  - Bonding and reactivity for oxygen, nitrogen and methane (C)
- Li adsorbed on SiC/Si or SiC; SiO<sub>2</sub>/Si and Si<sub>3</sub>N<sub>4</sub>/Si
  - Thermal treatment, same as for Si
  - Bonding and reactivity for oxygen and nitrogen
- Could be important information for choice of anode material

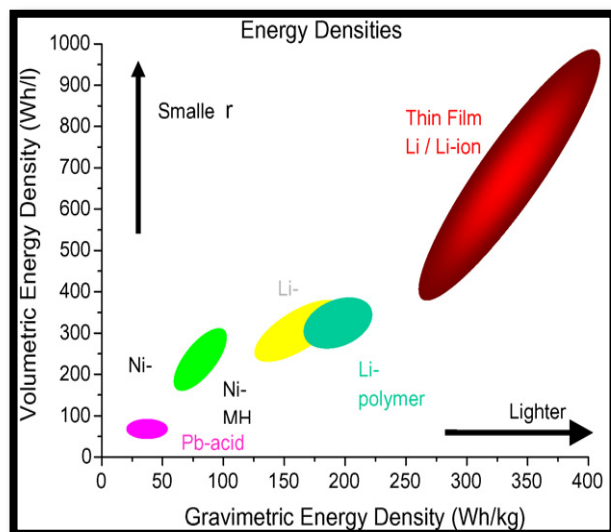
# Li-ion thin-film microbatteries



**G.MOHAN RAO**

**Plasma processing laboratory  
Dept.of Instrumentation and Applied Physics  
Indian Institute of Science, Bangalore**

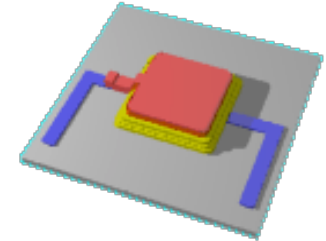
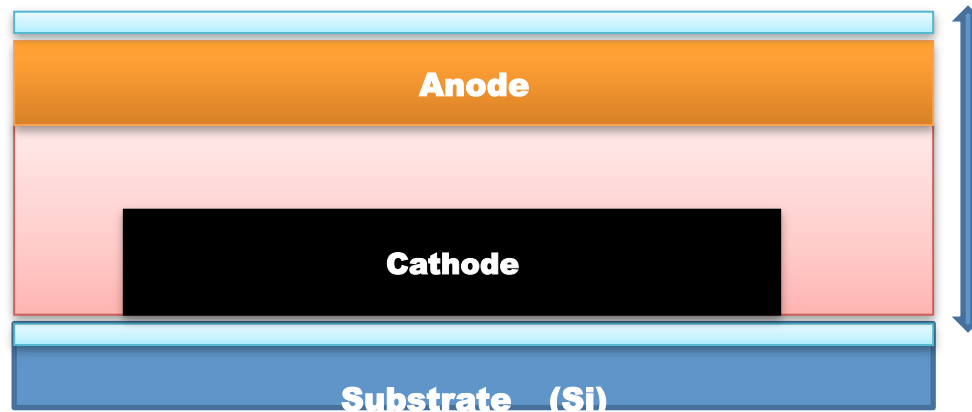
# Thin film batteries (Why?)



**$\text{LiCoO}_2$  | LiPON solid electrolyte | Li /  $\text{Sn}(\text{O})_2$  /  $\text{V}_2\text{O}_5$**

Battery Type	Voltage (V)	Energy Density		Discharge time (hrs)		(min) 1mm
		(Wh/kg)	(Wh/l)	5mm	1mm	
Ni-Cd	1.2	40	100	11.25	0.090	5.4
Ni-MH	1.2	90	245	27.60	0.221	13.2
Ag-Zn	1.5	110	220	24.75	0.198	11.9
Li-ion	3.6	155	400	45.00	0.360	21.6
Li-polymer	3.6	180	380	42.75	0.342	20.4
Thin Film Li-ion	3.6	250	1000	112.5	0.900	54

# Example of a thin film battery



Less than 10  $\mu\text{m}$

All solid state components

Thin film fabrication of all battery components

**Substrates**

**Si,  $\text{Al}_2\text{O}_3$ , quartz**

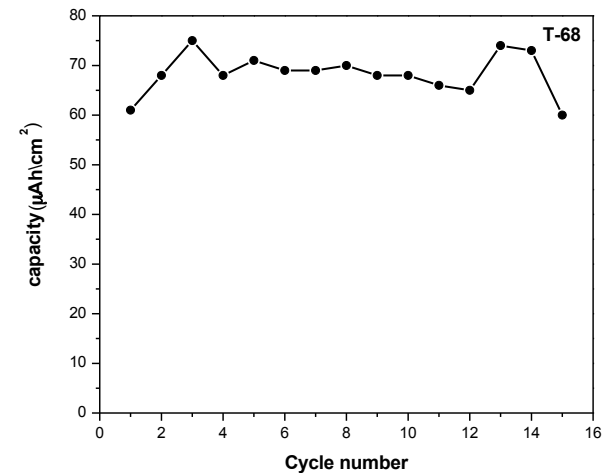
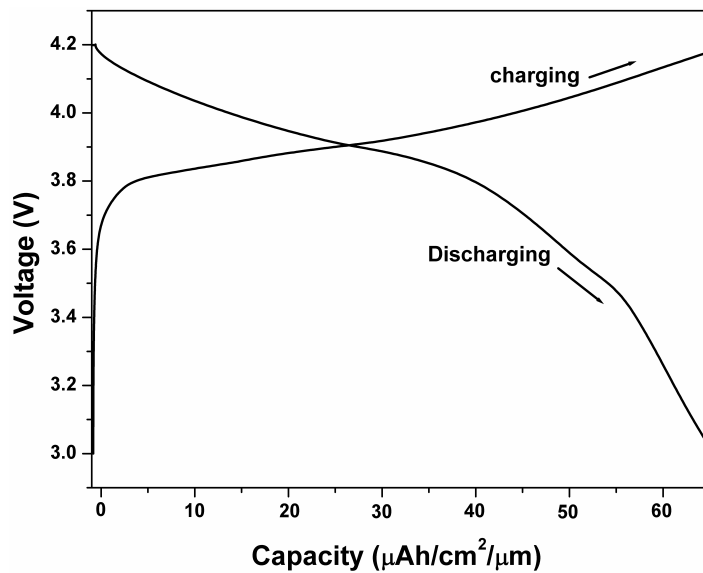
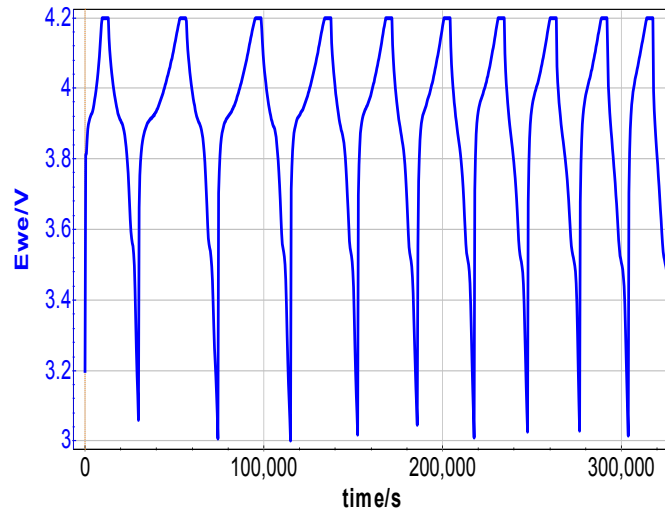
**Flexible substrate: Kapton**

# Realization of TFB - *Materials and processing*

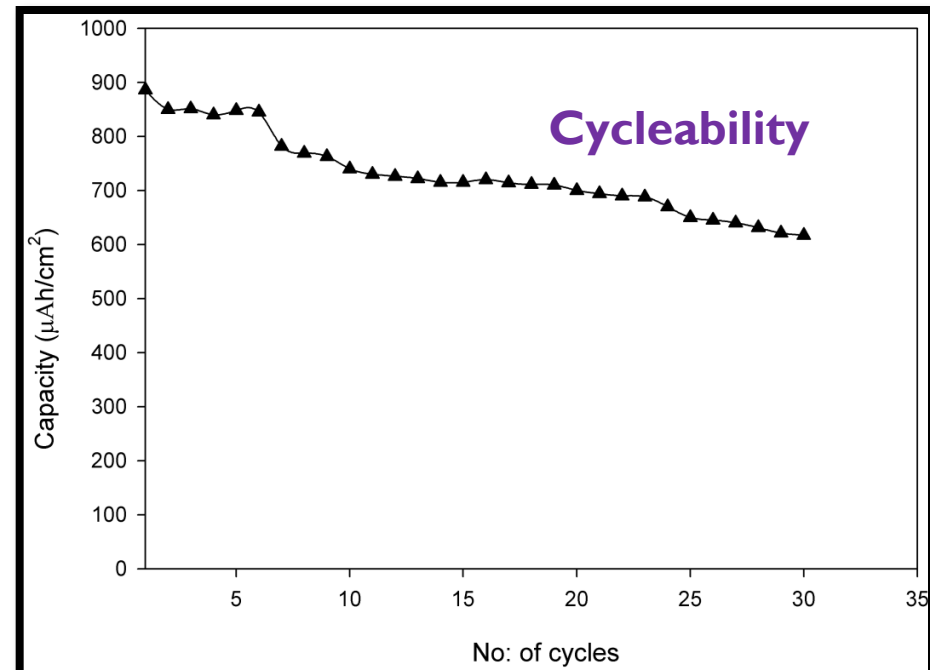
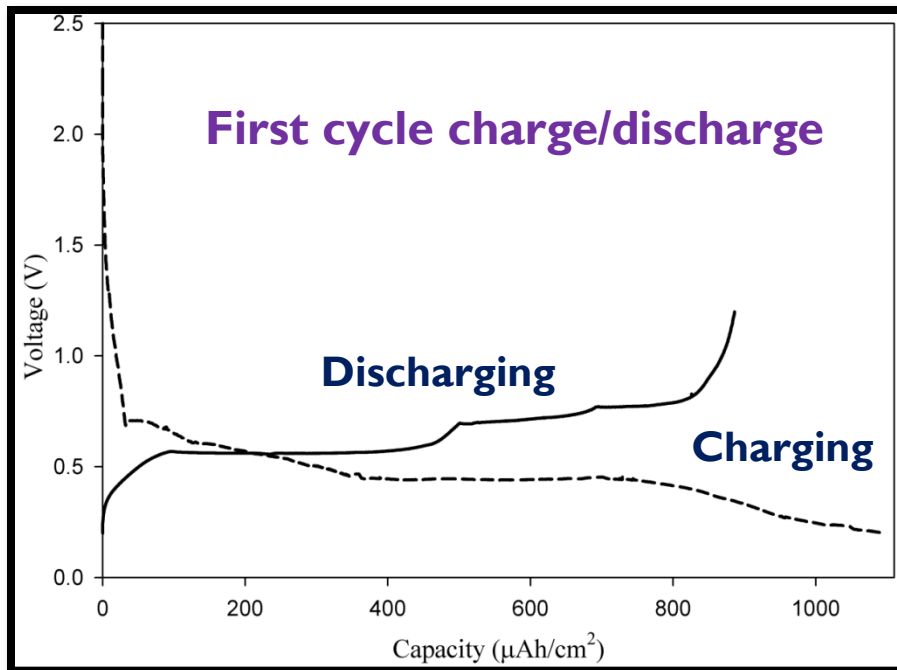
Layer	Material	Deposition Tech:
Cathode	$\text{LiCoO}_2$	RF sputtering in Ar and $\text{O}_2$ ambient. Post annealing at $700^\circ\text{C}$ for 2 hour.
Electrolyte	LiPON ( $\text{Li}_{2.9}\text{PO}_{3.3}\text{N}_{0.36}$ )	RF sputtering in $\text{N}_2$ ambient. Room temperature deposition Amorphous thin film is formed.
Anode	Metals (Cu,Sn,Al)	Sputtering / Evaporation TFB in discharged state Elemental Li can be avoided

**Substrate materials: Pt/Si, Stainless steel,  $\text{Pt/Al}_2\text{O}_3$**

# Electrochemical Performance of $\text{LiCoO}_2$ thin film



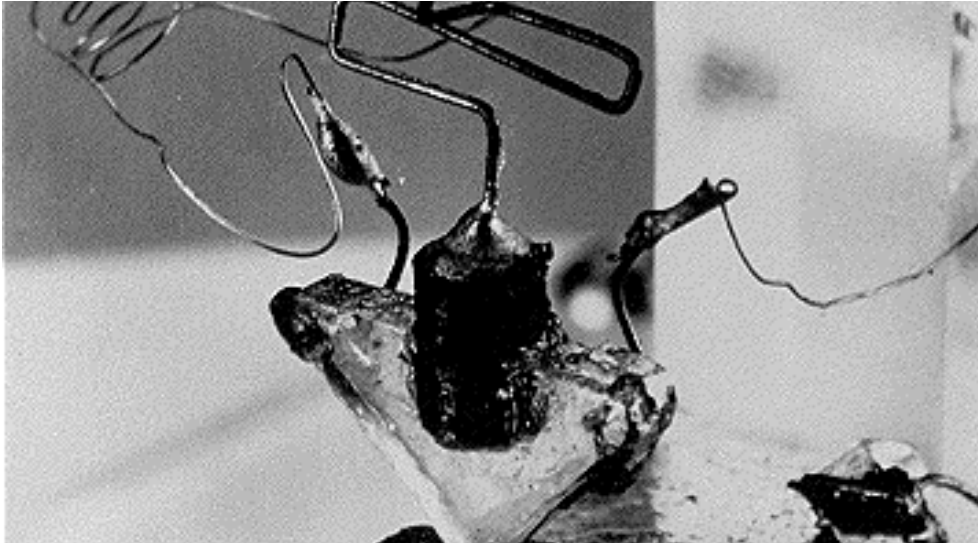
# Prototype TFB



- ❖ 'Nanobeads' of  $\approx 40\text{nm}$  size increases surface roughness
- ❖ volume change associated is taken care of due to porous nature
- ❖ capacity reduction up to 30 % after 30 cycles (compares well)



# From here...



# to here



**and here ?**



**Thank you**